



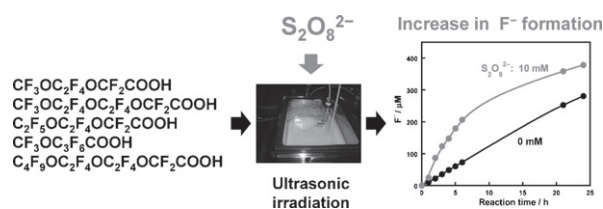
## Graphical Abstracts/J. Fluorine Chem. 141 (2012) 1–4

## Efficient decomposition of perfluoroether carboxylic acids in water with a combination of persulfate oxidant and ultrasonic irradiation

Hisao Hori<sup>ab</sup>, Yuta Nagano<sup>a</sup>, Misako Murayama<sup>b</sup>, Kazuhide Koike<sup>b</sup>, Shuzo Kutsuna<sup>b</sup><sup>a</sup>Department of Chemistry, Faculty of Science, Kanagawa University, 2946 Tsuchiya, Hiratsuka 259-1293, Japan<sup>b</sup>National Institute of Advanced Industrial Science and Technology (AIST), AIST Tsukuba West, 16-1 Onogawa, Tsukuba 305-8569, Japan

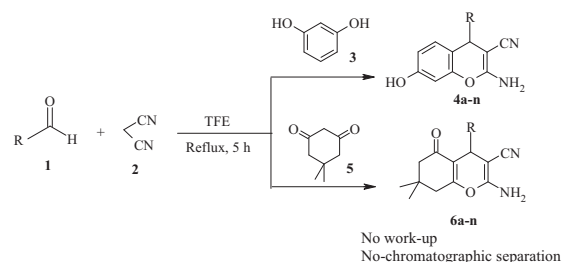
► Perfluoroether carboxylic acids were decomposed by  $S_2O_8^{2-}$  and ultrasonic irradiation. ► Perfluoroether carboxylic acids are alternatives to bioaccumulative surfactants. ►  $S_2O_8^{2-}$  enhanced the decomposition rates 2.5–3.9 times those without  $S_2O_8^{2-}$ .

J. Fluorine Chem., 141 (2012) 5

A facile and efficient synthesis of 2-amino-3-cyano-4*H*-chromenes and tetrahydrobenzo[*b*]pyrans using 2,2,2-trifluoroethanol as a metal-free and reusable mediumSamad Khaksar, Ahmad Rouhollahpour, Saeed Mohammadzadeh Talesh  
Department of Chemistry, Ayatollah Amoli Branch, Islamic Azad University, Amol, Iran

► In this study we examine the TFE as a new recyclable medium. ► 2-Amino-4*H*-chromene and tetrahydrobenzo[*b*]pyran were synthesized in trifluoroethanol. ► This method has the ability to tolerate a wide variety of substitutions. ► TFE was easily recovered.

J. Fluorine Chem., 141 (2012) 11

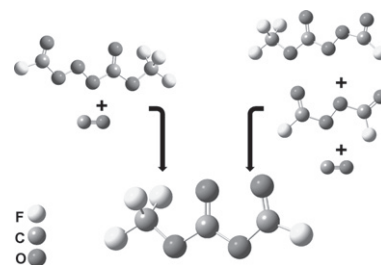
Synthesis and properties of trifluoromethoxyl fluoroformyl anhydride,  $CF_3OC(O)OC(O)F$ 

Martín M. Manetti, Gustavo A. Argüello, Maxi A. Burgos Paci

Instituto de Investigaciones en Físico Química de Córdoba (INFIQC) CONICET-UNC, Departamento, de Físico Química, Facultad de Ciencias Químicas, Universidad Nacional de Córdoba, Ciudad, Universitaria, X5000HUA Córdoba, Argentina

► The synthesis, IR spectroscopy and DFT calculations of the new perfluorinated anhydride,  $CF_3OC(O)OC(O)F$ , are presented. ► The new molecule completes the family of the asymmetric oxygen bonded acyl compounds  $CF_3OC(O)OxCOF$  WITH  $x = 1-3$ . ► A mechanism explaining the formation of the target molecule from the combination of  $CF_3OC(O)Ox$  and  $FCOx$  radicals is discussed.

J. Fluorine Chem., 141 (2012) 16



## Novel anionic fluorine-containing amphiphilic self-assembly polymer micelles for potential application in protein drug carrier

Guoqiang Liu<sup>a</sup>, Wen Fan<sup>a</sup>, Ling Li<sup>a</sup>, Paul K. Chu<sup>b</sup>, Kelvin W.K. Yeung<sup>c</sup>, Shuilin Wu<sup>abc</sup>, Zushun Xu<sup>abc</sup>

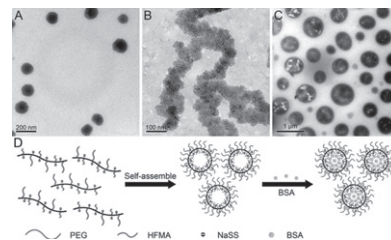
<sup>a</sup>Ministry-of-Education Key Laboratory for the Green Preparation and Application of Functional Materials, Hubei University, Wuhan 430062, China

<sup>b</sup>Department of Physics & Materials Science, City University of Hong Kong, Tat Chee Avenue, Kowloon, Hong Kong, China

<sup>c</sup>Division of Spine Surgery, Department of Orthopaedics and Traumatology, The University of Hong Kong, Pokfulam, Hong Kong, China

► We prepared anionic fluorine-containing amphiphilic graft polymers. ► The copolymers could self-assemble into core-shell structure micelles. ► There was a visible adsorption between the micelles and BSA. ► The micelles have potential application in protein drug carrier.

*J. Fluorine Chem.*, 141 (2012) 21



## Removal of fluoride ions from aqueous solutions using protonated cross-linked chitosan particles

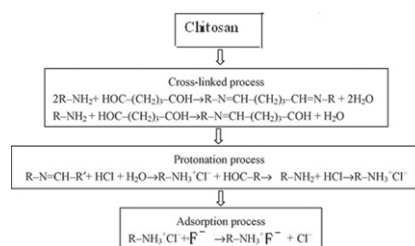
Ruihua Huang<sup>a</sup>, Bingchao Yang<sup>b</sup>, Qian Liu<sup>a</sup>, Kelian Ding<sup>a</sup>

<sup>a</sup>College of Science, Northwest A&F University, Yangling, Shaanxi 712100, China

<sup>b</sup>Xi'an Institute of Geology and Mineral Resource, Xi'an, Shaanxi 710054, China

► PCP was prepared with chitosan membrane cross-linked and protonated. ► This adsorbent has good subsidence. ► The maximum amount of fluoride adsorbed was  $8.10 \text{ mg g}^{-1}$  at  $20^\circ\text{C}$ . ► The equilibrium data fitted with Langmuir and Freundlich isotherm models well. ► The adsorption of fluoride onto this adsorbent was attributed to physical adsorption.

*J. Fluorine Chem.*, 141 (2012) 29

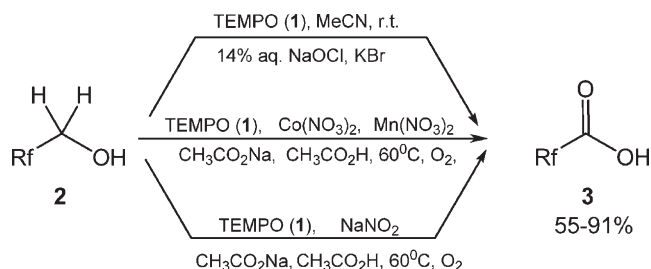


## TEMPO mediated oxidation of fluorinated alcohols to carboxylic acids

Jolanta Ignatowska<sup>a</sup>, Oleg Shyshkov<sup>b</sup>, Tilman Zippies<sup>b</sup>, Klaus Hintzer<sup>b</sup>, Gerd-Volker Röscenthaler<sup>a</sup>

<sup>a</sup>School of Engineering and Science, Jacobs University Bremen GmbH, Campus Ring 1, D-28759 Bremen, Germany

<sup>b</sup>Dyneon GmbH, D-84504 Burgkirchen Werk Gendorf, Germany



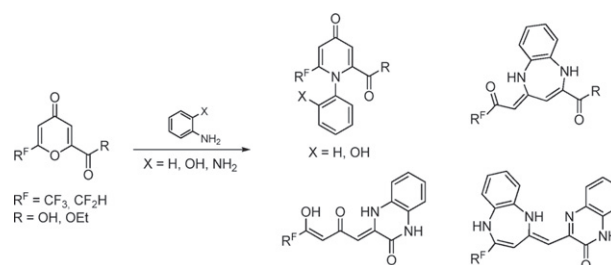
*J. Fluorine Chem.*, 141 (2012) 35

## Reactions of 6-(tri- and 6-(difluoromethyl))comanic acids and their ethyl esters with aniline and its 2-substituted derivatives

Dmitrii L. Obydenov, Boris I. Usachev

Department of Chemistry, Ural Federal University, Kuybysheva st., 48, 620026 Ekaterinburg, Russia

► Reactions of 6-(tri- and 6-(difluoromethyl))comanic acids with aniline and its 2-substituted derivatives were investigated. ► R<sup>F</sup>-containing 4-pyridones, benzodiazepines and quinoxalinones were synthesized. ► Electrophilic properties of 6-CF<sub>3</sub>-comanic acid were evaluated.



*J. Fluorine Chem.*, 141 (2012) 41

## Sphingosine and clavaminol H derivatives bearing fluorinated chains and their cytotoxic activity

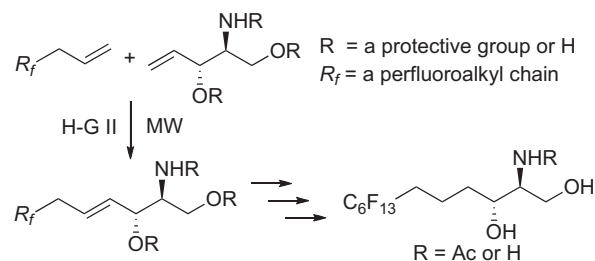
*J. Fluorine Chem.*, 141 (2012) 49

Eva Prchalová<sup>ab</sup>, Ivan Votruba<sup>b</sup>, Martin Kotora<sup>ab</sup>

<sup>a</sup>Department of Organic and Nuclear Chemistry, Faculty of Science, Charles University in Prague, Hlavova 8, 128 43 Praha 2, Czech Republic

<sup>b</sup>Institute of Organic Chemistry and Biochemistry AS CR, v.v.i., Flemingovo nám. 2, 166 10 Praha 6, Czech Republic

- Cross-metathesis of perfluoroalkylpropenes with functionalized alkenes.
- Synthesis of fluorinated sphingosine derivatives.
- Synthesis of fluorinated clavaminol H analogues.
- Biological activity of the prepared fluorinated clavaminol H analogues.



## Reaction of 2-(trifluoromethyl)chromones with pyridoxal: Formation of 1-benzopyranooxepino- and 1-benzopyranopyranopyridines

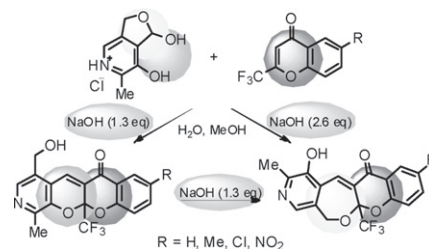
*J. Fluorine Chem.*, 141 (2012) 58

Vyacheslav Ya Sosnovskikh<sup>a</sup>, Vladislav Yu Korotaev<sup>a</sup>, Alexey Yu Barkov<sup>a</sup>, Anna A. Sokovnina<sup>a</sup>, Mikhail I. Kodess<sup>b</sup>

<sup>a</sup>Department of Chemistry, Ural Federal University, pr. Lenina 51, 620083 Ekaterinburg, Russian Federation

<sup>b</sup>Institute of Organic Synthesis, Ural Branch of the Russian Academy of Sciences, 620041 Ekaterinburg, Russian Federation

- 2-(Trifluoromethyl)chromones and pyridoxal hydrochloride were reacted in the presence of sodium hydroxide.
- Two novel polyfluoroalkyl-containing heterocyclic systems, 11a, 13-dihydro-6H-1-benzopyrano[3',2':6,7]oxepino[3,4-c]pyridin-6-ones and 6H,11aH-1-benzopyrano[3',2':5,6]pyrano[2,3-c]pyridin-6-ones were synthesized.
- The reaction depends on the pH of the aqueous media.



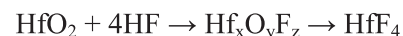
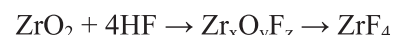
## A thermogravimetric study of the fluorination of zirconium and hafnium oxides with anhydrous hydrogen fluoride gas

*J. Fluorine Chem.*, 141 (2012) 64

Bernard M. Vilakazi, Oduetse S. Monnahela, Jacobus B. Wagener, Pieter A.B. Carstens, Tshupo Ntsoane

Research and Development Division, South African Nuclear Energy Corporation Limited, P.O. Box 582, Pretoria 0001, South Africa

- Dynamic reaction of  $\text{ZrO}_2/\text{HfO}_2$  with hydrogen fluoride.
- Isothermal reactions of  $\text{ZrO}_2/\text{HfO}_2$  with hydrogen fluoride.
- Characterization of the products of the reactions of  $\text{ZrO}_2/\text{HfO}_2$  with HF and confirmation of the tetrafluoride through XRD.



## Multifunctional surface modification of an aramid fabric via direct fluorination

*J. Fluorine Chem.*, 141 (2012) 69

Euigyung Jeong<sup>a</sup>, Bum Hoon Lee<sup>b</sup>, Song Jun Doh<sup>c</sup>, In Jun Park<sup>d</sup>, Young-Seak Lee<sup>a</sup>

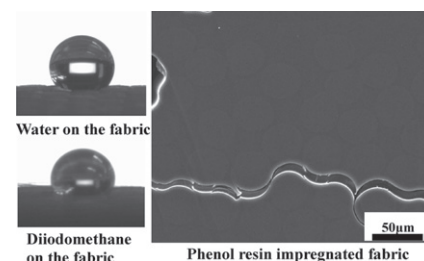
<sup>a</sup>Department of Fine Chemical Engineering and Applied Chemistry, BK21-E<sup>2</sup>M, Chungnam National University, Daejeon 305-764, Republic of Korea

<sup>b</sup>Heracron Research Institute, Kolon Industries, Inc., 212 Gongdan-dong, Gumi-Si, Gyung-sangbuk-do 730-030, Republic of Korea

<sup>c</sup>Department of Textile Convergence of Biotechnology & Nanotechnology, Convergence Technology R&D Division, Korea Institute of Industrial Technology, 1271-18 Sa-3-dong, Sangrok-gu, Ansan-si Gyeonggi-do 426-791, Republic of Korea

<sup>d</sup>Research Center for Biorefinery, Korea Research Institute of Chemical Technology, Daejeon 305-600, Republic of Korea

- Direct fluorination of the aramid fabric improved hydrophobicity and oleophobicity.
- Direct fluorination of the aramid fabric improved phenol resin wettability and impregnation.
- These improvements suggest that direct fluorination of the aramid fabric can be an efficient multifunctional surface modification method.



## Synthesis of *N*-substituted $\alpha,\alpha$ -difluoro- $\beta$ -aminophosphonates by addition of diethyl lithiodifluoromethylphosphonate to imines

Prabhakar Cherkupally, Petr Beier

Institute of Organic Chemistry and Biochemistry,  
Academy of Sciences of the Czech Republic,  
Flemingovo nám. 2, 166 10 Prague, Czech Republic

► Diethyl lithiodifluoromethylphosphonate adds to *N*-substituted imines to provide *N*-substituted  $\alpha,\alpha$ -difluoro- $\beta$ -aminophosphonates.  
► Reactions with *N*-aryl or *N*-alkyl aldimines provide high yields of products. ► In ketimine series only activated *N*-(2,2,2-trifluoro-1-phenylethylidene)aniline showed high reactivity.

